

# Conceivable Bottlenecks of RRR300 Nb Production for ILC Project

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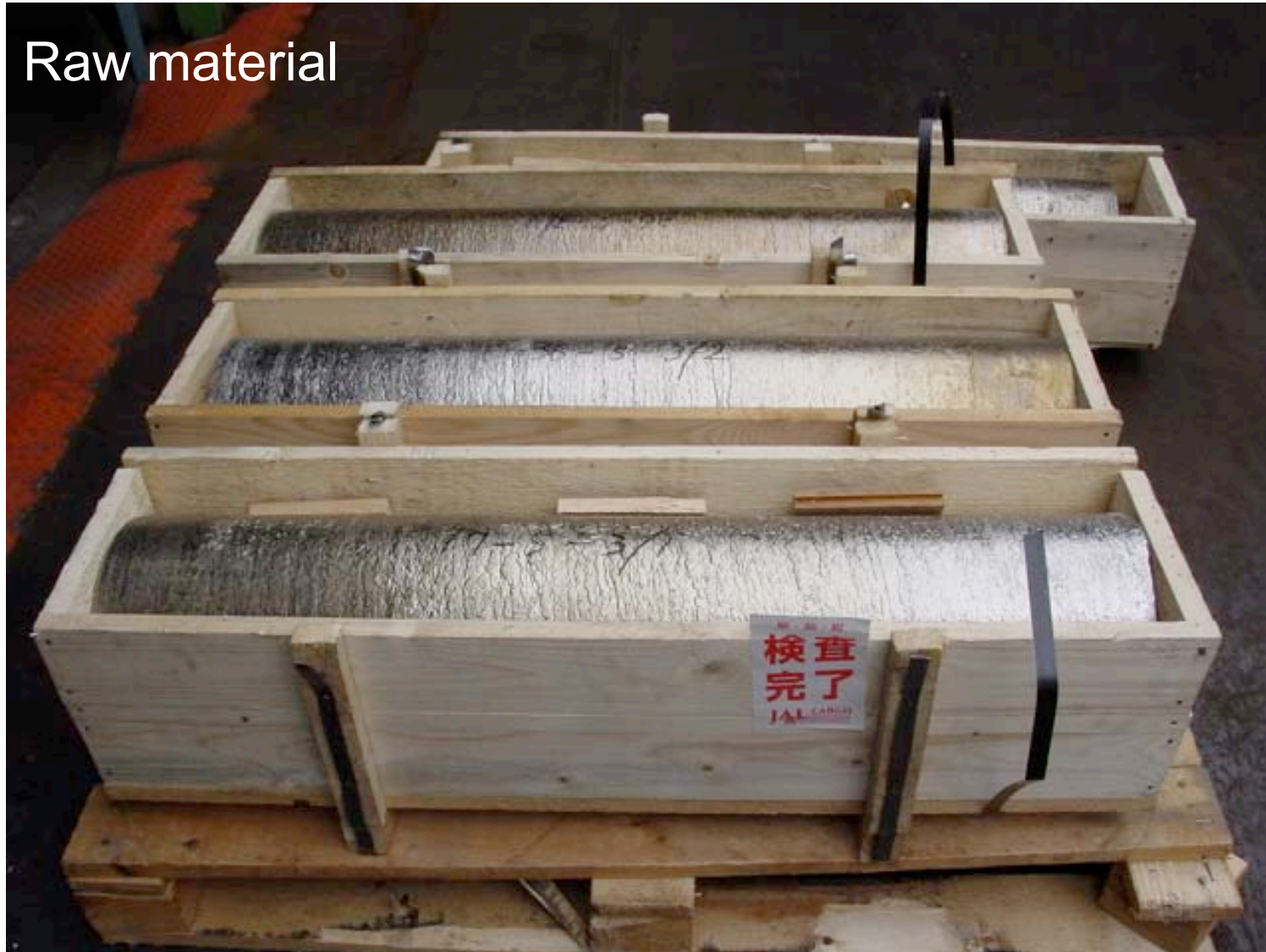
# Agenda

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1. How to produce high purity niobium.
2. Conceivable bottlenecks of production for huge amount of high purity Nb.

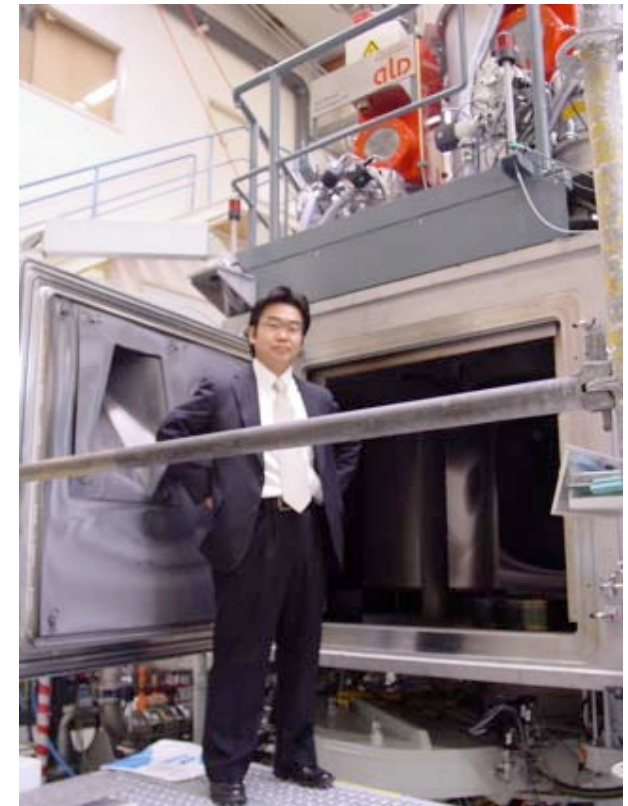
# Niobium Ingots

Raw material



# Electron Beam Melting Furnace

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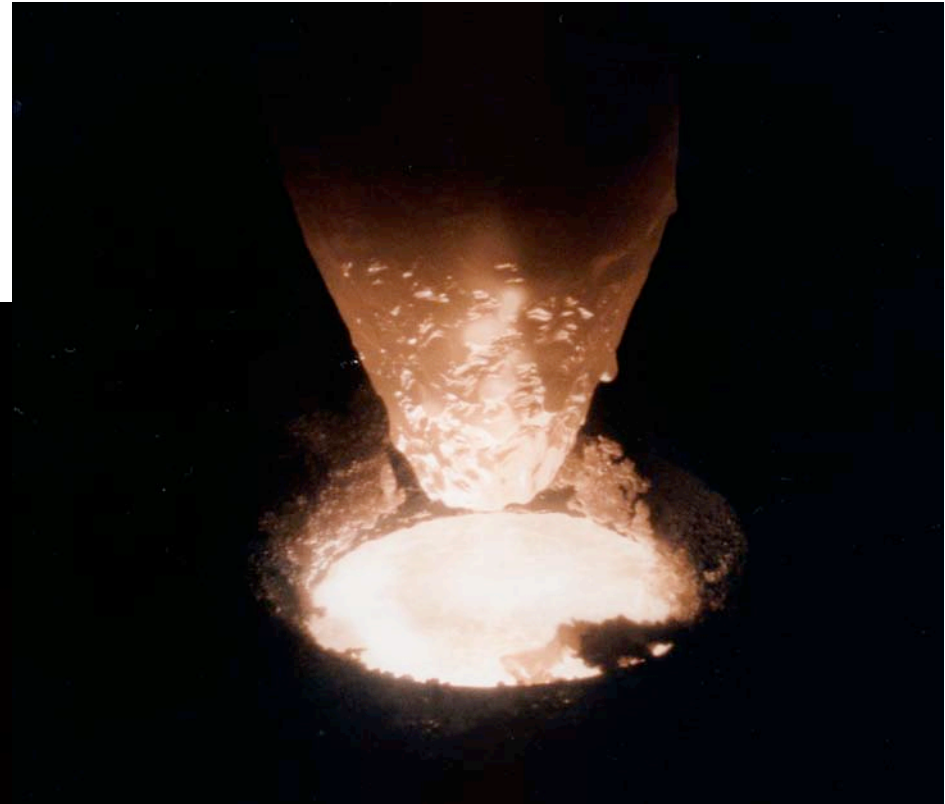


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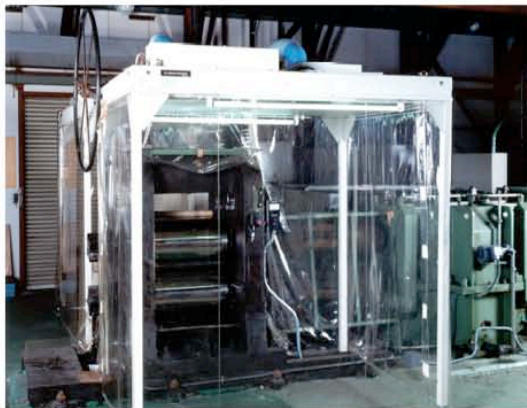
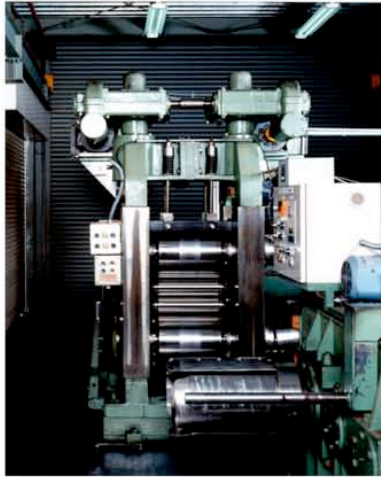


# EB Melting

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# Rolling mills



Name	Roll Size		Motor	Specific Sheet Size	
	Diameter	Width		Thickness	Width
Two High Rougher Mill	500 mm $\varnothing$	1,000 mm	300 kW	60–8 mm	
Two High Finishing Mill	500 mm $\varnothing$	1,100 mm		8–0.8 mm	900 mm
Two High Rougher Mill	460 mm $\varnothing$	800 mm	75 kW	8–3 mm	
Two High Finishing Mill	420 mm $\varnothing$	800 mm		3–0.1 mm	500 mm
Four High Mill	390/130 mm $\varnothing$	450 mm	37 kW	1–0.1 mm	350 mm
Four High Mill	300/145 mm $\varnothing$	300 mm	37 kW	0.3–0.08 mm	200 mm
Six High Mill	40/105 mm $\varnothing$	150 mm	5.5 kW	0.3–0.05 mm	100 mm

# Vacuum heat treatment

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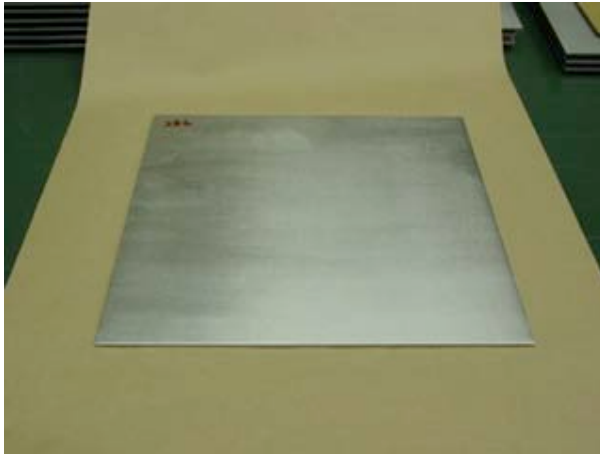


Maximum temperature: 1400 [°C]  
Achievable pressure:  $2 \times 10^{-5}$  [Pa]



# Niobium sheets

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Nb sheets  
used for producing  
Superconducting cavity material  
at  
Fermi Lab SMTF



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# Demand of high purity Nb for ILC project

# Superconducting RF cavity for ILC

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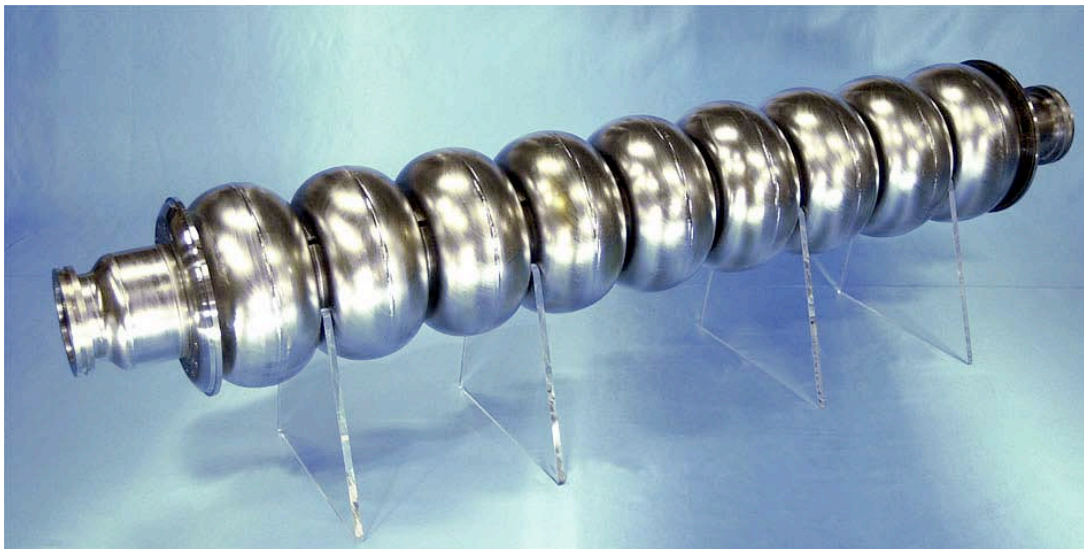
17,000 nine cell cavities

17,000 x 18 half-cells = 306,000 sheets

Nb weight of a sheet for half-cell:  $2.8 \times 265 \times 265 \text{ mm} = 1,685 \text{ g}$

$306,000 \times 1.685 \text{ kg/sheet} = 515.6 \text{ ton}$

Backup cavity, beam tube, HOM coupler, etc...about 600 ton

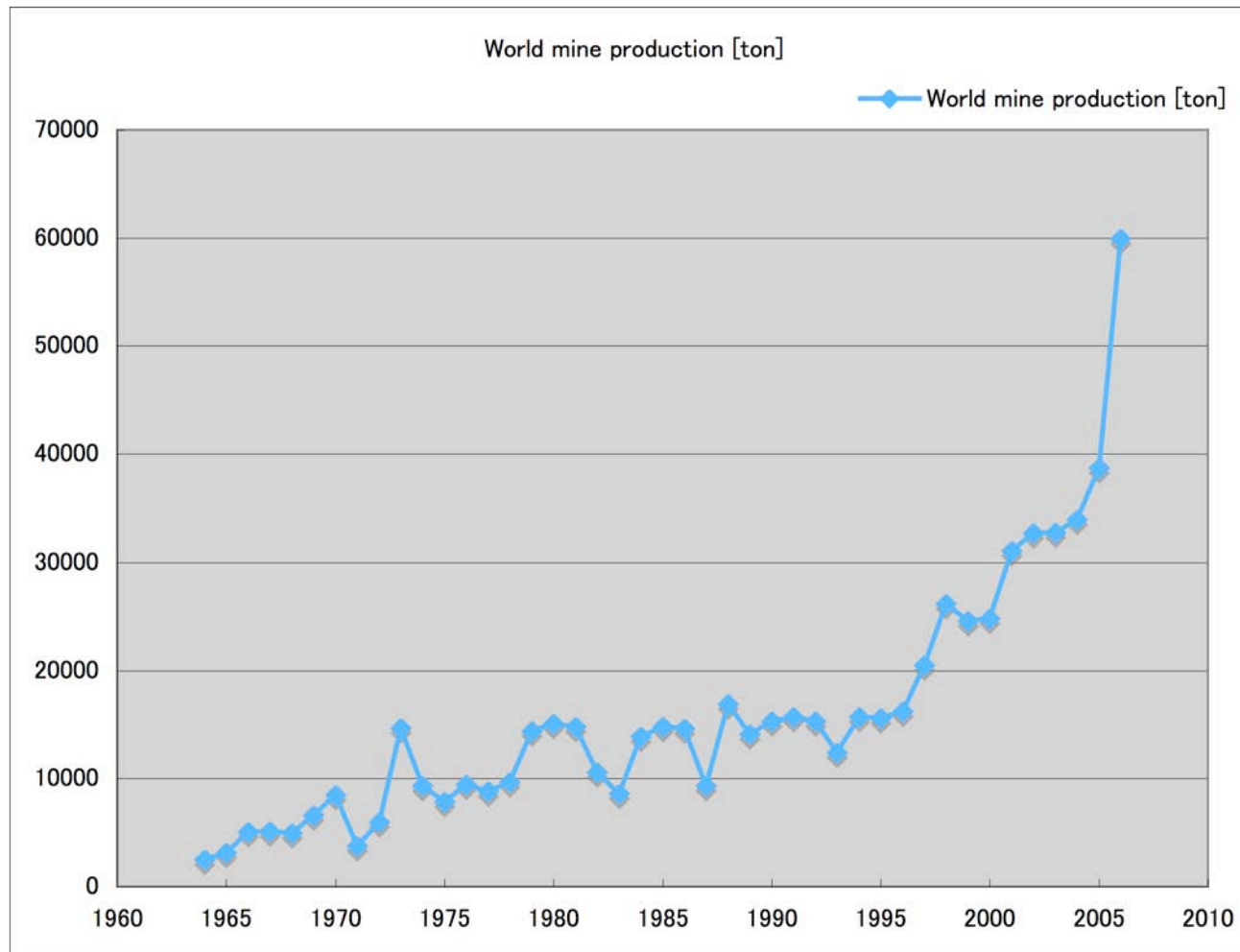


<http://lcdev.kek.jp/photos/20050513/20050513lchiCav.jpg>

ILC 9-cell Cavity made with  
high purity Nb  
17,000 cavities / 3 years

Demand of High purity Nb is  
600 ton in 3 years.

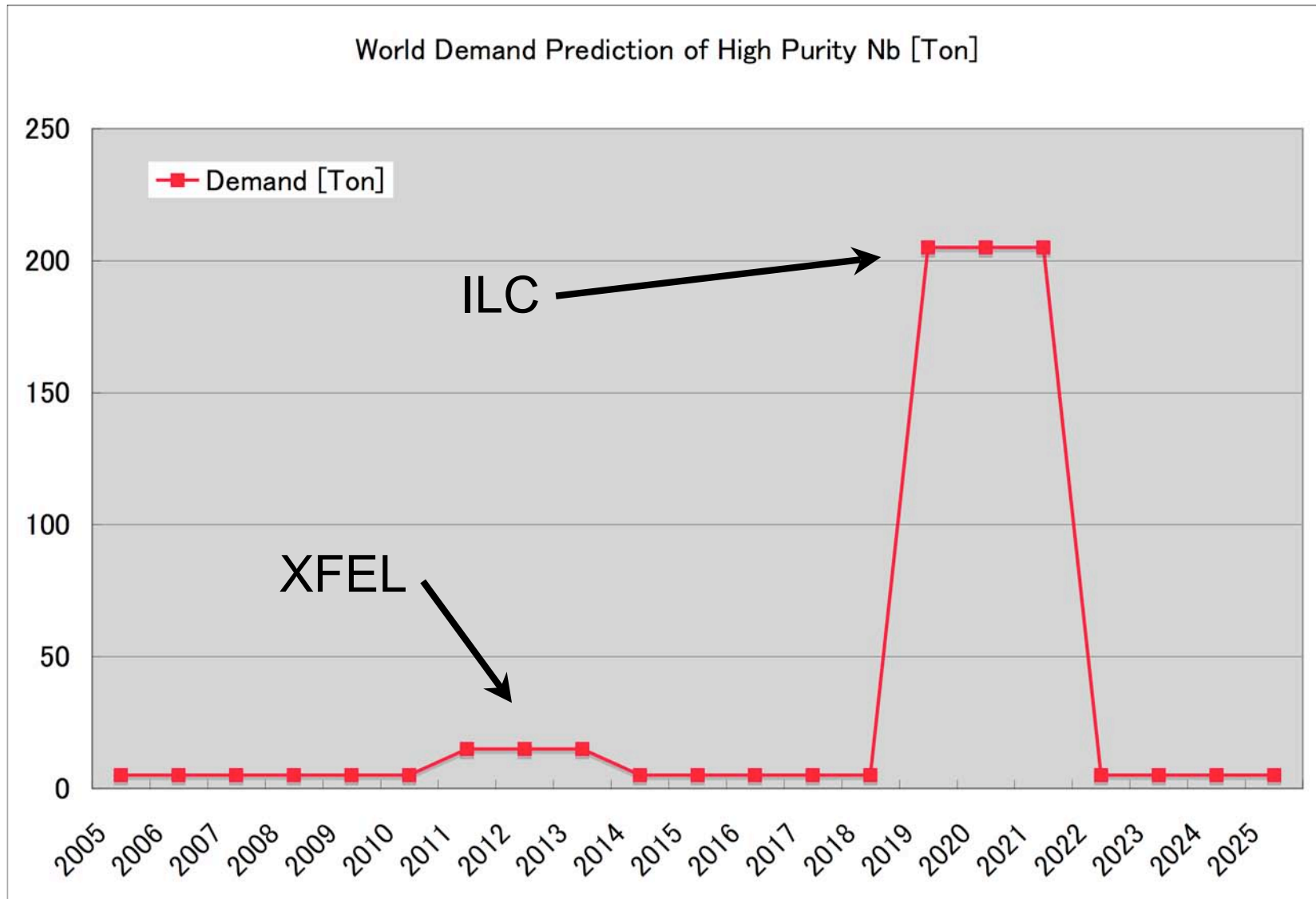
# Niobium world mine production



U.S. Geological Survey, Mineral Commodity Summaries 1996–2007



# Demand Forecast

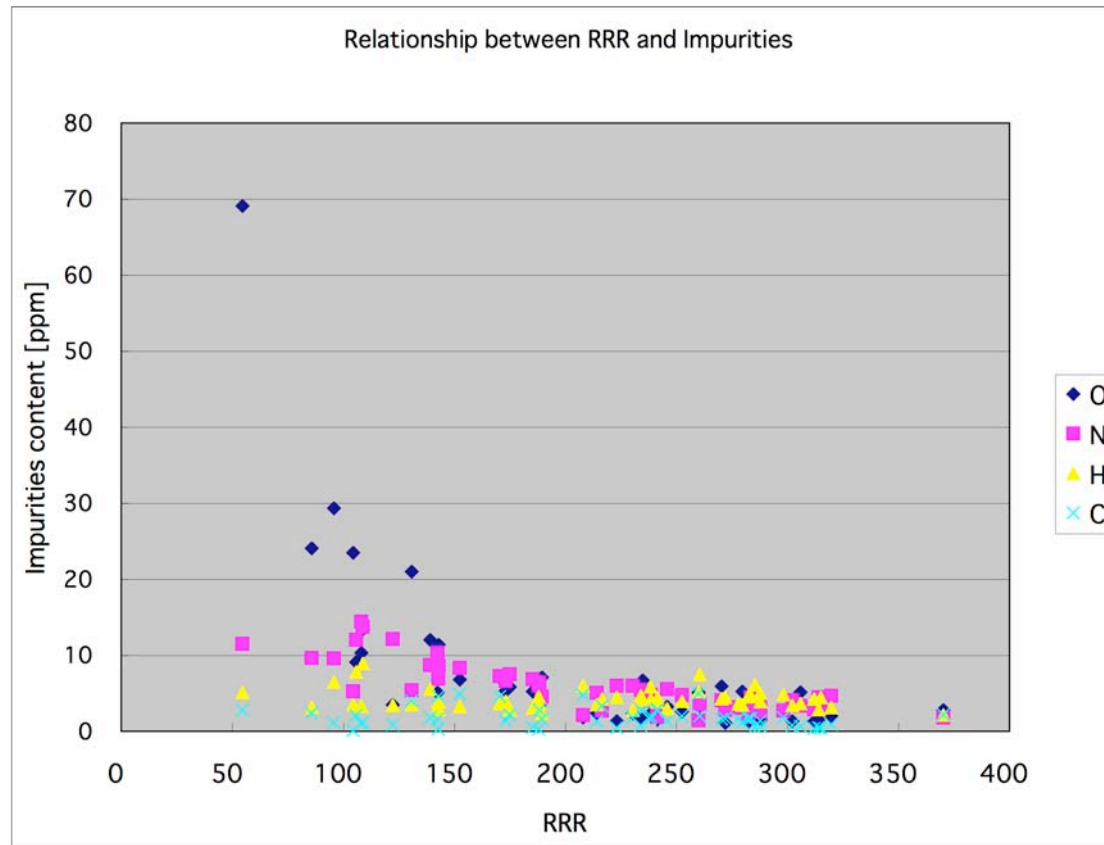


# Bottleneck 1

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- Low Ta concentrate Nb Raw materials

# Relationship between RRR and Interstitial impurities





# Regression Analysis Result

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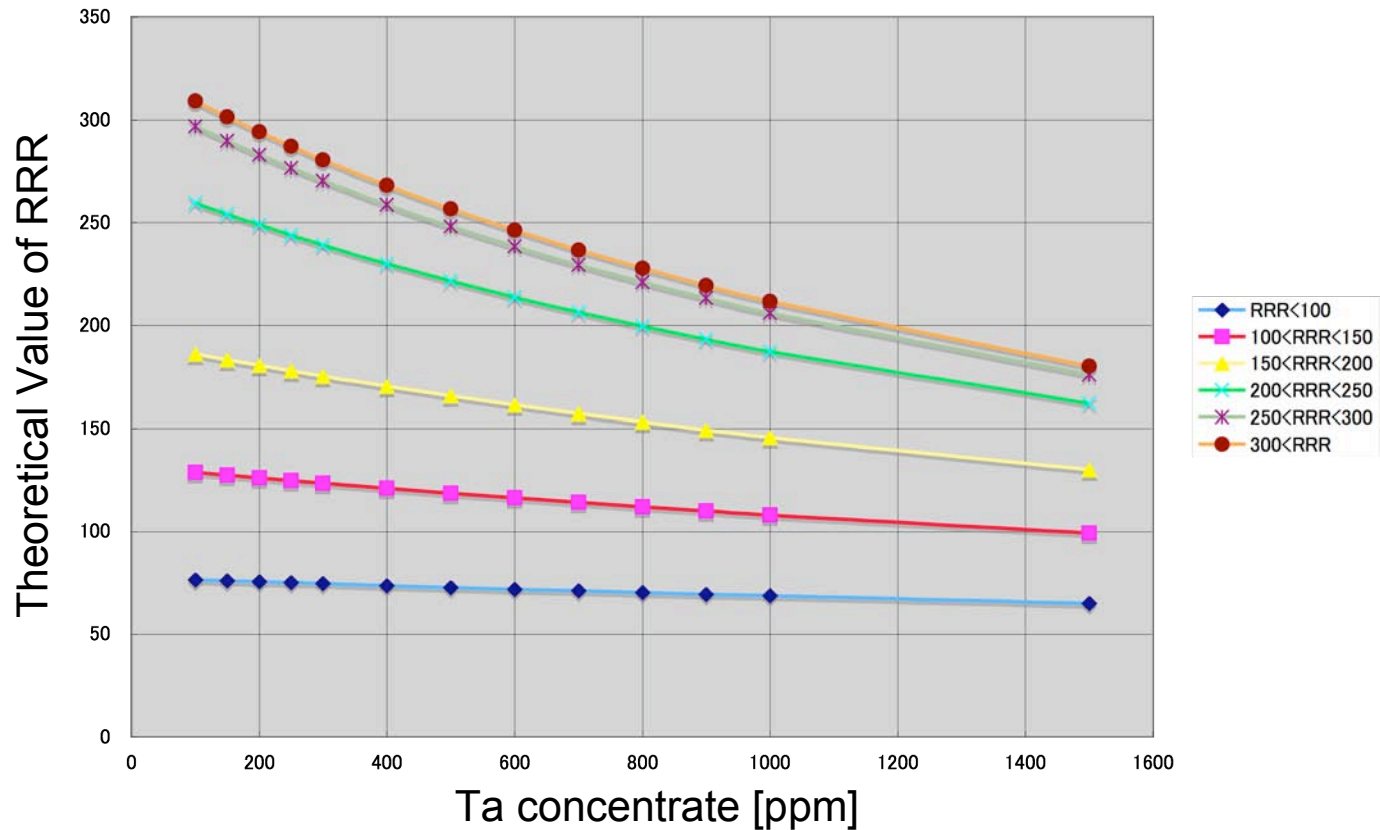
K.K.Schulze: J. Metals, 33(1981), 33-41

$$\frac{1}{RRR} = \frac{O}{5000} + \frac{N}{3900} + \frac{H}{1550} + \frac{C}{4100} + \frac{Ta}{550000} + \dots$$

Umezawa's calculation.

$$\frac{1}{RRR} = \frac{O}{5800} + \frac{N}{2273} + \frac{H}{16322} + \frac{C}{8911} + \frac{Ta}{604690} + \frac{1}{1249}$$

# Simulation of Ta influence



$$\frac{1}{RRR} = \frac{O}{5800} + \frac{N}{2273} + \frac{H}{16322} + \frac{C}{8911} + \frac{Ta}{604690} + \frac{1}{1249}$$

# F. Schoelz et al., "How to produce NbRRR~600 on an industrial scale" Proc. of the 9th Workshop on RF Superconductivity, Santa Fe, USA, Nov. 1-5, 1999, pp. 100-103

## How to produce Nb<sub>RRR-600</sub> on an industrial scale

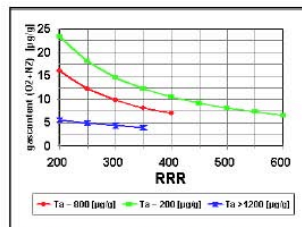
Friedhold Schoelz, W.C. HERAEUS GMBH & CO KG, MATERIALS TECHNOLOGY DIVISION, HERAEUSSTRASSE 12-14, 63450 HANAU, GERMANY

### 1. Introduction

To produce high purity Niobium (RRR  $\geq 600$ ) on an industrial scale, some points have to be considered. One of the factors that limits the RRR-values, in addition to the gas content, is the residual Ta content of the starting-material: the Ta content enriches during electron beam-melting and forms very stable carbides and nitrides, which makes the degassing process very difficult. Thus it is absolutely necessary to use a starting material with a low Ta content. Additionally it is possible to have a better degassing of the niobium by using appropriate changes during the melting process. On the other hand it is very important to exclude all steps during processing, which return the gases or carbon to the material, e.g. welding etc. Also new getter-materials were tested during the last treatment like Ca. To control the very purity, it was necessary to create a special high precise referencematerial. This material was developed in cooperation with BAM Berlin and the working group for analytical chemistry of GdCh in Germany. Finally, it is also necessary to control the semifinished products like sheets or tubes before starting the expensive processing, to prevent failure of the finished cavities. That is the reason, why we have used Squids that are very sensitive and have a high resolution.

### 2. Correlation between RRR, the gas content and the Ta content

During the last year we made an extensive calculation based on 400 measured RRR samples and integrated them with a multiple regression over the Ta



Correlation between RRR, gas, and Ta content

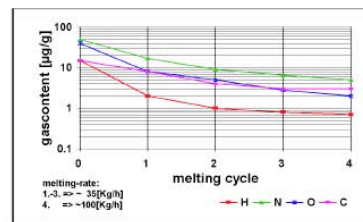
content and the gas content. We discovered that RRR values  $>400$  could only be reached by a Ta-content lower than  $200 \mu\text{g/g}$ . The best method to produce such a raw material on an industrial scale is the liquid-liquid-extraction with MIBK or MIPK as shown in the following description:

$\text{NbCl}_5 + \text{TaCl}_5 + \text{MIBK} / \text{MIPK} + \text{H}_2\text{O}$   
Ta in the aqueous phase  
Nb in the organic phase

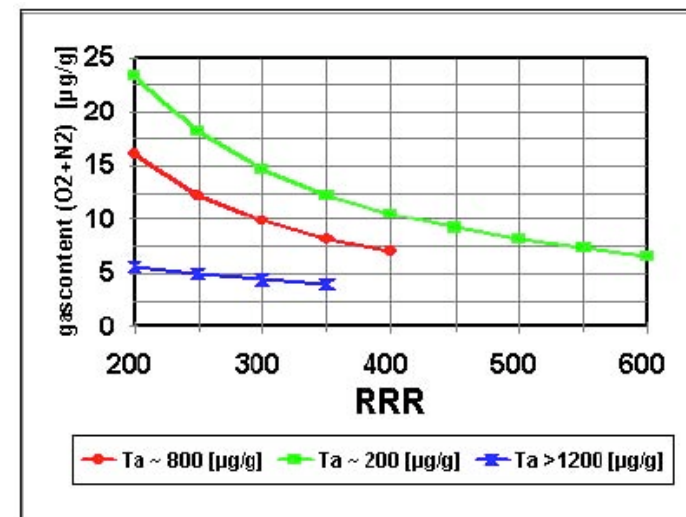
$\text{K}_2\text{NbOF}_7 + \text{K}_2\text{TaF}_7 + \text{MIBK} / \text{MIPK} + \text{H}_2\text{O}$   
Ta in the organic phase  
Nb in the aqueous phase

After a cascade of 20 extractions the Ta content is down to about  $200 \mu\text{g/g}$ . The theoretical value after 36 extractions is  $< 20 \mu\text{g/g}$ . So we produced more than 2 [to] of Nb with a Ta content of  $200 \mu\text{g/g}$  in the standard sheet process and reached a RRR value better than 450 at once.

### 3. Degasing process during EB melting



The purification and degassing process during electron beam melting is very complex. All metals with a lower melting point than Nb will be evaporated. Metals with a higher melting point like Ta and W will be enriched in the melt, because Nb also evaporates as Nb-oxides during the melting process. All gases are continuously in exchange with the hot metal surface. Gases from the furnace atmosphere will at first be adsorped on the metal surface and then dissolved in the metal or they produce evaporable products. Dissolved gases in the metal will diffuse to the surface and produce molecular

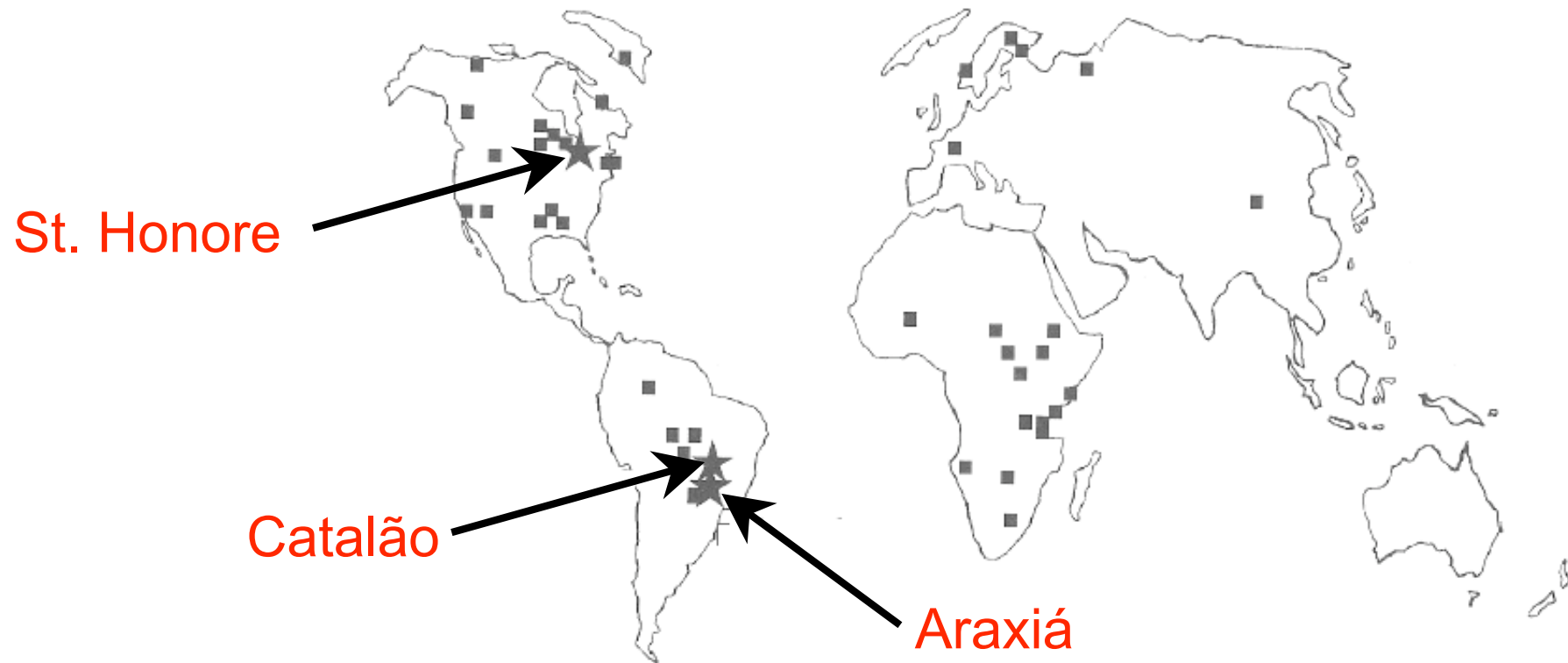


Correlation between RRR, gas, and Ta content



# Nb mines

- **Columbite**  $(\text{Fe, Mn})(\text{Nb, Ta})_2\text{O}_6$
- **Tantalite**  $(\text{Fe, Mn})(\text{Nb, Ta})_2\text{O}_6$
- **Pyrochlore**  $(\text{Ca, Na})_2\text{Nb}_2\text{O}_6\text{F}$



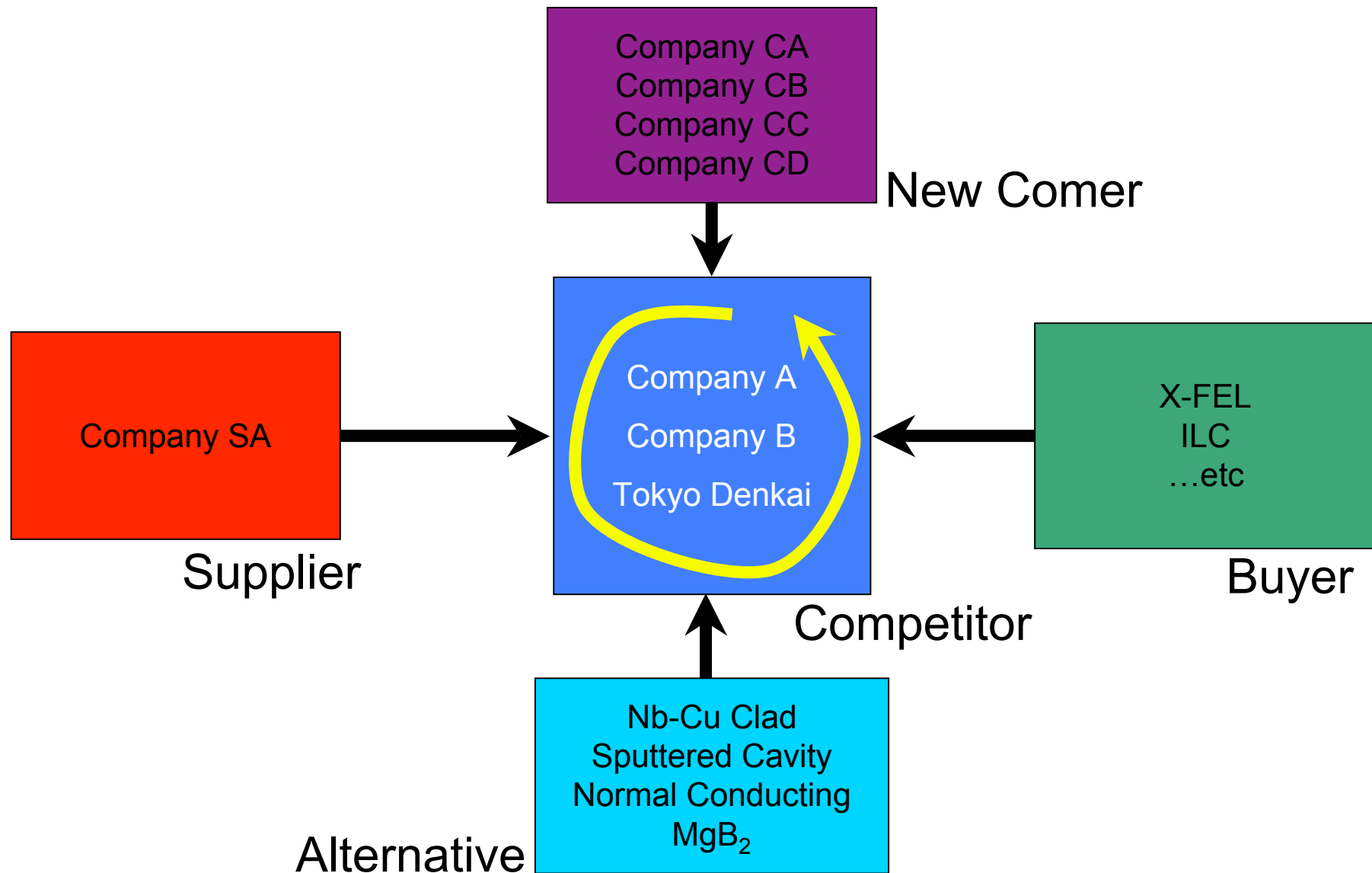
Imagumbai, et al: "Resource, production volume & manufacturing technology of niobium" Material Science & Technology Vol.72(2002)No.3 pp204-210

# Bottleneck 2

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Melting capacity

# Five Force Analysis of Nb for SC Cavities

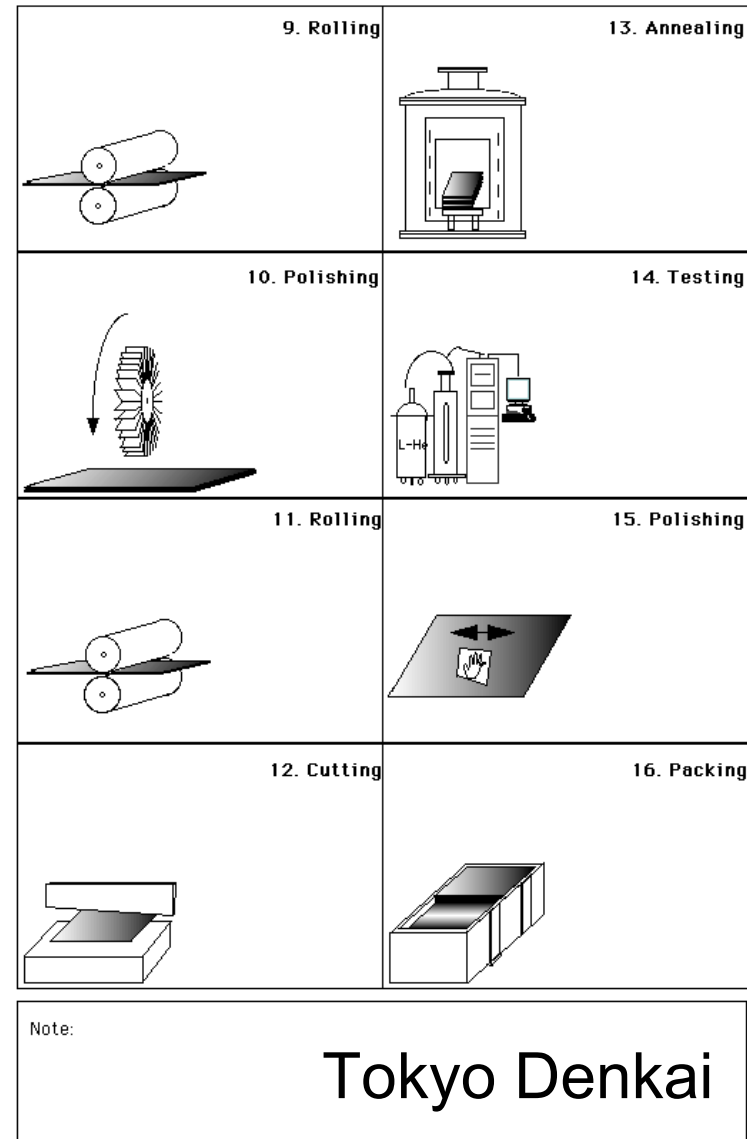
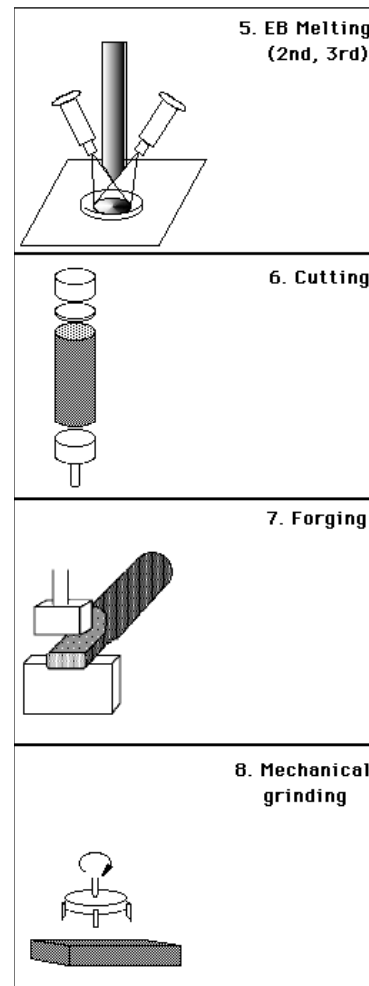


# Bottleneck 3

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Sheet production process

# Sheet production process





# Multi Wire Saw Slicing



# Summary

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- **Small market**
  - The demand for high purity Nb for superconducting accelerator is several tons per year, although the total Nb demand is 60 thousand tons per year.
  - Only three vendors are qualified for the XFEL project.
  - These three vendors have several main businesses. It can be said that it is difficult to concentrate the high purity Nb production even if only 3 years.
- **All Nb vendors are buying Nb ingots from the same company**
  - Producing high RRR Nb requires a raw material with a low Ta content. Today, only one company provides it.
- **Sheet production**
  - Sheet production is not easy. It needs experience. For example, material and shape of roll, timing of cross roll, intermediate annealing, grain size control, thickness irregularity...

# Summary

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- We have a time to consider to increase the high purity niobium production capacity.
- We would like to seek the solution.

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Thank you for attention!